## MARK SCHEME for the May/June 2012 question paper

## for the guidance of teachers

## 9701 CHEMISTRY

9701/51

Paper 5 (Planning, Analysis and Evaluation), maximum raw mark 30

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2012 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



Page 2	Mark Scheme: Teachers' version	Syllabus	Paper
	GCE AS/A LEVEL – May/June 2012	9701	51

Question	Sections	Indicative material	Mark
1 (a) (i)	PLAN Problem	States that the moles of copper(II) hydroxide increase as the molar concentration of copper(II) sulfate increases and sketches a line from the origin with an initial positive gradient. Ignore any subsequent plateau or maximum on this line.	[1]
(ii)		A straight line terminating at the point of saturation with marked co-ordinates: award 2 marks.	[1]
		A straight line clearly terminating within the grid but without marked co-ordinates: award 1 mark.	[1]
		A line (not necessarily straight) which does not terminate at the saturation point but with the co-ordinates marked: award 1 mark. This line can plateau after the saturation point.	
(b)	PLAN Problem	(i) concentration of copper(II) sulfate.	[1]
	Problem	(ii) moles of copper(II) hydroxide	[1]
(c)	PLAN Method	Indicates at least five experiments. These may be shown in the table in <b>1(e).</b> Five blank rows in the table are acceptable.	[1]
		A range of concentrations over at least 0.8 moldm <sup>-3</sup> , which must cover 1.0 moldm <sup>-3</sup> , up to a maximum of 1.39. Accept a range of mass of copper(II) sulfate (with solution volume) that has been calculated satisfying the same concentration criteria.	[1]
		Filtering/centrifuging	[1]
		Method of drying and weighing the precipitate. Include washing with water (and propanone), (air) drying and weighing (to constant mass). Do not accept direct heating, blotting or a statement that the precipitate is simply left to dry.	[1]
		A suitable calculation of a molarity, even if greater than 1.39 ( $M_r$ of the copper(II) sulfate must be used). Check that the solution is made up to the appropriate volume and not that a mass is added to a fixed volume of water.	[1]
(d)	PLAN Method	Identifies that copper(II) sulfate is harmful/a danger to the environment.	[1]
		Identifies that sodium hydroxide is corrosive (from the hazcard information).	[1]
		Give one mark for a precaution for either hazard of (chemical) resistant gloves or large dilution when disposing of chemicals.	[1]

Page 3	3	Mark Scheme: Teachers' version	Syllabus	Paper
		GCE AS/A LEVEL – May/June 2012	9701 51	51
(e)	PLAN Method	This table must match the plan in <b>1(c)</b> . Five required depending on whether serial dilut the solutions: Either (i) mass of copper(II) sulfate/g or (ii sulfate solution/cm <sup>3</sup> and volume of water/c mass/weight of copper(II) hydroxid concentration of copper(II) sulfate/ number of moles of copper(II) hydr The full word for the unit can be used with Ignore other column headings and units. If Five or Four are fully correct, two marks; one mark; otherwise zero.	ion is used to pre ) volume of copp m <sup>3</sup> for serial dilu e/g; mol.dm <sup>-3</sup> ; oxide (no unit). or without / or ()	epare ber tions; [2]
				[Total: 15]

	Page 4		Mark Scheme: Teachers' version	Syllabus	Pape	r
		(	GCE AS/A LEVEL – May/June 2012	9701	51	
2	(a)	ACE Data	The required two column headings temper and volume, (B + 26), /cm <sup>3</sup> are fully correct unit can be used with or without / or ().			[1]
			Both columns are fully completed to the co significant figures. Allow 2 errors.	rrect number of		[1]
	(b)	ACE Data	Label the <i>x</i> -axis temperature and the <i>y</i> -axis unambiguous label is acceptable e.g. colur must be present in the accepted forms. The cover at least half the grid in both direction be on the given grid. If a true origin has be candidate's line is extrapolated back to the be included as a "plotted point" in the asse half the grid in both directions. This mark n plots.	nn label (A). United points in s and all points in en scaled in and re then the origin ssment of cover	its must must the n is to ing	[1]
			Check the plotting of points 1, 3, 8, 10 and	any obvious erro	or.	[1]*
			Give one mark for drawing a 'straight-line of here from incorrectly plotted points).	of best fit' (allow	'ecf'	[1]*
	(c)	ACE Evaluation	Incorrect plots of <b>(i)</b> volume/temp °C and <b>(i</b> change/temp will still allow these marks to other wrong plots will not. Allow the candidate to select up to five and include that furthest from the line.	be accessed. Ar		
			All the anomalous points are circled on the unambiguously stated in the text.	grid or		[1]
			For each of the two different anomalies an explanation gains one mark.	appropriate		[2]
			Point 3 (V low). Gas not equilibrated with the temperature, OR volume read before all the <u>increased</u> temperature. OR volume read be <u>increased</u> temperature.	e gas attained th		
			Point 8 (V high). Gas not equilibrated with temperature, OR volume read before all the <u>decreased</u> temperature. OR volume read before all the <u>decreased</u> temperature.	e gas attained th		
			If the candidate suggests that the gas is no anomalous points but does not specify the temperature change to that point, award 1	direction of	or both	

Page 5			aper 51
			•
<b>、</b> /	vCE ata	(i) For, two pairs of construction lines on the graph drawn from the line to the axes and for correctly deducing the graph values of these two intercepts, give one mark. There has to be some indication on the graph of the intercepts used to calculate the gradient. These could be table points, provided they are indicated on the graph. If the true origin has been used in calculating the slope then only one pair of lines and one intercept is necessary.	t [
		A correctly calculated value of the slope using the candidate's figures. The mark is for the magnitude (ignore units). If the candidate used the true origin in the slope calculation then two zeros are not needed in the calculation.	[
		If the slope expression is inverted, then the calculation mark is lost but the intercept value mark can be gained.	
		(ii) For a correctly read intercept at 273K give one mark. This mark is available for other permitted wrong plots as in 2(c), but for a volume change/temp plot 26 cm <sup>3</sup> has to be added to the read intercept. Allow a calculated volume only if the candidates line on the graph was extrapolated to the true origin. Units must be included.	[
· · /	CE Evaluation	The data is reliable as most (6 or more) of the points/results/data lie on the line of best fit. Accept few anomalous points.	[
		If there are 6 or more points not on the line then accept that the results are not reliable.	
()	CE Conclusion	These marks not available for other plots.	
		For a statement that the 'law' is justified because a straight line (direct proportionality) is obtained give one mark.	[
		The data confirms the relationship $V = kT$ or that volume is directly proportional to absolute temperature. Calculations showing that V/T is the same for more than 1 point on the graph are worth 1 mark.	[
(3)	CE Conclusion	A second line on the grid above the original line. This second line to have a greater slope (not parallel) and not touching the origina line unless at the true origin (if used). Again, this mark is available for the other permitted plots as in <b>2(c)</b> .	
		(* is mark available for other plots)	
		·	otal: