MARK SCHEME for the May/June 2013 series

9701 CHEMISTRY

9701/32

Paper 32 (Advanced Practical Skills 2), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2013 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



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Question Sections		Sections	Indicative material		Mark	
1 (a) MMO Collection		MMO Collection	I All thermometer readings and two weighings + correct mass of FB 1 recorded.	1		
		PDO Recording	II Correct headings and units in table (lines not needed). Weighings and mass to same no. dp. <i>Must use solidus, brackets or describe unit fully in words. If units not included in column headings, every entry must have the correct unit shown.</i>	1		
			III Temperature recorded to 0.5 °C (must have at least one ending in .5 °C and one ending in .0 ℃).	1		
		MMO Quality	Award IV and V if $\delta \le 0.30 ^{\circ}\text{C g}^{-1}$ Award IV if $0.30 < \delta \le 0.60 ^{\circ}\text{C g}^{-1}$		[5]	
1	(b)	PDO Layout	 Uniform scales chosen to use more than half of each axis. At least 5 large squares on <i>y</i>-axis. Axes labelled, units not needed. Scale extends at least 1 °C below lowest recorded point. 	1		
			II All points correctly plotted to within ½ small square and in correct small square.	1		
			III Appropriate lines of best fit drawn.	1		
		ACE Interpretation	IV Lines extrapolated and correct value (within 0 .5 °C) of ΔT from graph (ignore sf and sign).	1	[4]	
1	(c)	PDO Recording	 Table drawn to include weighings, correct mass of FB 2, initial and final thermometer readings (ignore units), headings must be unambiguous. 	1		
		MMO Quality	Award II and III if $\delta \le 0.30 \text{ °C g}^{-1}$ Award II if 0.30 < $\delta \le 0.60 \text{ °C g}^{-1}$	1 1	[3]	

Page 3					Syllabus	Paper	
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(d) (i) (iii) (ii)	Display		1	Shows 25 x $4.3 \times \Delta T$ for step 1 and for need to check calculation). Should be $\frac{answer \ to \ (i) \ x \ 84.0}{mass \ FB \ 1 \ x \ 1000}$	or step 2 (no	1	
(iv)			ш	Calculates $\frac{answer \ to \ (iii) \ x \ 106.0}{mass \ FB \ 2 \ x \ 1000}$		1	
						1	
(v)	ACE Cor	∃ iclusions		Award IV and V if correctly calculates answers to (ii) and (iv) with correct sid (answer to (iv) $- 2 \times$ answer to (ii) score (iv) $-$ (ii) scores 1 2(ii) $-$ (iv) scores 1)	gn	1	
	PD0 Disp		VI	Signs in (ii) (+ve) and (iv) (–ve).		1	
			VII	All final answers given to 2 or 4 sf (mi answers attempted).	nimum of 3	1	[7]
(e) (i)	ACE Inte	E rpretation	Single balance reading \pm 0.005 or 0.01 g for 2dp balance. $\Delta m \pm 0.01$ or 0.02 g for 2 dp balance. 0.05 or 0.1 g for 1 dp balance / 0.0005 or 0.001 for 3 dp.			1	
(ii)		<i>rea</i> Cal moi dp l	mber of dps must correspond to candic dings. culates max Δm error / mass FB 2 in S re sf (2 minimum for 1 dp balance and palance). ww ecf.	step 2 to 2 or	1	[2]	
(f) (i)	ACE Imp	E rovements	volu	dent incorrect as acid already in exces ume gives smaller ΔT /greater volume r it energy and/or greater % error.	•	1	[1]
(ii)	MM Inte	O rpretation		empts to calculate $\Delta T / m$ or $m / \Delta T$ for atios of m_1/m_2 and T_1 / T_2 or similar (no).		1	
	MM Dec	0 cisions	Cor	rect calculation (1.35 / 1.37 or 0.74 / 0 nclusion – yes, values concordant / cor ues differ.		1	[2]
			1			[Tot	al: 24
						[Tota	ai: 24

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	F	B 4 MnC <i>l</i> ₂	$\mathbf{FB 5 Fe}(NO_3)_3 + HNO_3 \qquad \mathbf{FB 6 ZnSO_4}$	₁ + H₂S0	D 4
2	(a)	MMO Decisions	I Selects any 2 from barium chloride or nitrate, silver nitrate or lead nitrate.	1	
			II NaOH (allow OH if ion formulae already penalised) + Al and warm for NO ₃ ⁻ .	1	
			For III and IV any two from		
		MMO Collection	 only FB 4 gives white ppt with Ag⁺ 	1	
			 only FB 6 gives white ppt with Ba²⁺ 	1 1	[5]
		 FB 4 and FB 6 give a white ppt with Pb²⁺ (allow relevant dash as alternative to ppt) 		[0]	
			(ppt in any other test with Ag ⁺ , Pb ²⁺ or Ba ²⁺ is a CON)		
			${\bf V}$ FB 5 only forms gas/ NH_3 turns litmus paper blue		
2 (b)	(b)	MMO Collection	FB 4 gives off-white / buff / light brown ppt insoluble in both and darkening of the ppt in either.	1	
			FB 5 gives red-brown/brown/rust/orange-brown (not red, not orange) ppt insoluble in excess of both.	1	
			FB 6 gives white ppt soluble in excess of both	1	[3]
			or Award 1 mark if all NaOH or NH ₃ observations, including in excess, are correct except 'darkening of ppt in (i) .		
2	(c)	ACE	FB 4 contains Mn^{2+} and Cl^{-} .	1	
		Conclusions	FB 5 contains Fe^{3+} and NO_3^{-} .	1 1[3]	
			FB 6 contains Zn^{2+} and SO_4^{2-} .		
2 (0	(d)	MMO Decisions	 Selects suitable tests: two of Mg / Zn / Fe / Na₂CO₃, litmus paper/U.I. paper, chromate. (specific names of reagents not needed) (expected results not needed) 	1	
		PDO Layout	II Single table (no repeat headings)(must be the 2 reagents stated above).	1	
		MMO Collection	III Correct observations for FB 4 and either FB 5 or FB 6. For FB 4 correct observation is allowed if no +ve test for gas.	1	
			IV Correct observations for other from FB 5 and FB 6 .	1	
		ACE Conclusions	V FB 5 and FB 6 contain H ⁺ .	1	[5]
		1		1	l: 16